



# ABG Interpretation

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# Overview

- ◆ An ABG analysis evaluates how effectively the lungs are delivering oxygen to the blood and how efficiently they are eliminating carbon dioxide from it.
- ◆ The test also indicates how well the lungs and kidneys are interacting to maintain normal blood pH (acid-base balance).

# Overview cont.

- ◆ ABG's are usually done to assess:
- ◆ respiratory disease and other conditions that may affect the lungs, and to manage patients receiving oxygen therapy (respiratory therapy)
- ◆ In addition, the acid-base component of the test provides information on kidney function too.

# Indications

- ◆ Respiratory failure - in acute and chronic states.
- ◆ Severe illness leading to metabolic acidosis including; cardiac, liver and renal failure, burns, multi-organ failure, sepsis, hyperglycemia, toxins.
- ◆ Acute Asthma  $SaO_2 < 92\%$
- ◆ Ventilated patients
- ◆ Patients involved in sleep studies
- ◆ Any severely unwell patients.

# Contraindications

- ◆ AV fistula (dialysis)
- ◆ Negative Allen's Test
- ◆ Fractured wrist
- ◆ Poor peripheral circulation (PVD)

# Procedure

- ◆ Direct arterial puncture usually radial artery, alternatives include femoral and brachial in emergency settings and the ulnar and dorsalis pedis.
- ◆ Essential to ensure collateral flow (Allen's Test)
- ◆ Explain the procedure to the patient - it is painful.
- ◆ If there is time then local anaesthesia can be used.
- ◆ ABG syringes usually come prepacked and are heparinised. Some contain a vacuum and thus the plunger does not always need to be pulled.

# Procedure cont.

- ◆ The wrist is extended - a pillow under the hand may improve comfort.
- ◆ Palpate the artery and hold fingers firmly over the pulsation.
- ◆ Then introduce the needle at a  $45^\circ$  angle slowly with the bevel facing upwards, aiming for the point of maximum pulsation.
- ◆ Once you hit the artery, try to obtain at least a 1 ml sample.
- ◆ Once you have taken your sample and withdrawn the needle, apply firm pressure for a minimum of 2 minutes

# Procedure Cont.




- ◆ If multiple samples are required then an indwelling arterial cannula can be placed.
- ◆ Allow the patient to titrate with the oxygen for 5-10 minutes (30 minutes if they have chronic obstructive pulmonary disease (COPD)) before taking a sample.
- ◆ Remember ICE!!!!



# Interpretation of Results

- ◆ Normal pH is 7.35-7.45
- ◆ Value  $<7.35$  is acidotic
- ◆ Value  $>7.45$  is alkalotic
- ◆ Acidosis & Alkalosis can be caused by a problem with the respiratory system or a metabolic cause
- ◆ Can also have combined respiratory/metabolic states

# Is it Respiratory or Metabolic?

1. Respiratory Acidosis       Increased pCO<sub>2</sub> >50
2. Respiratory Alkalosis       Decreased pCO<sub>2</sub> <30
3. Metabolic Acidosis       Decreased HCO<sub>3</sub> <18
4. Metabolic Alkalosis       Increased HCO<sub>3</sub> >30

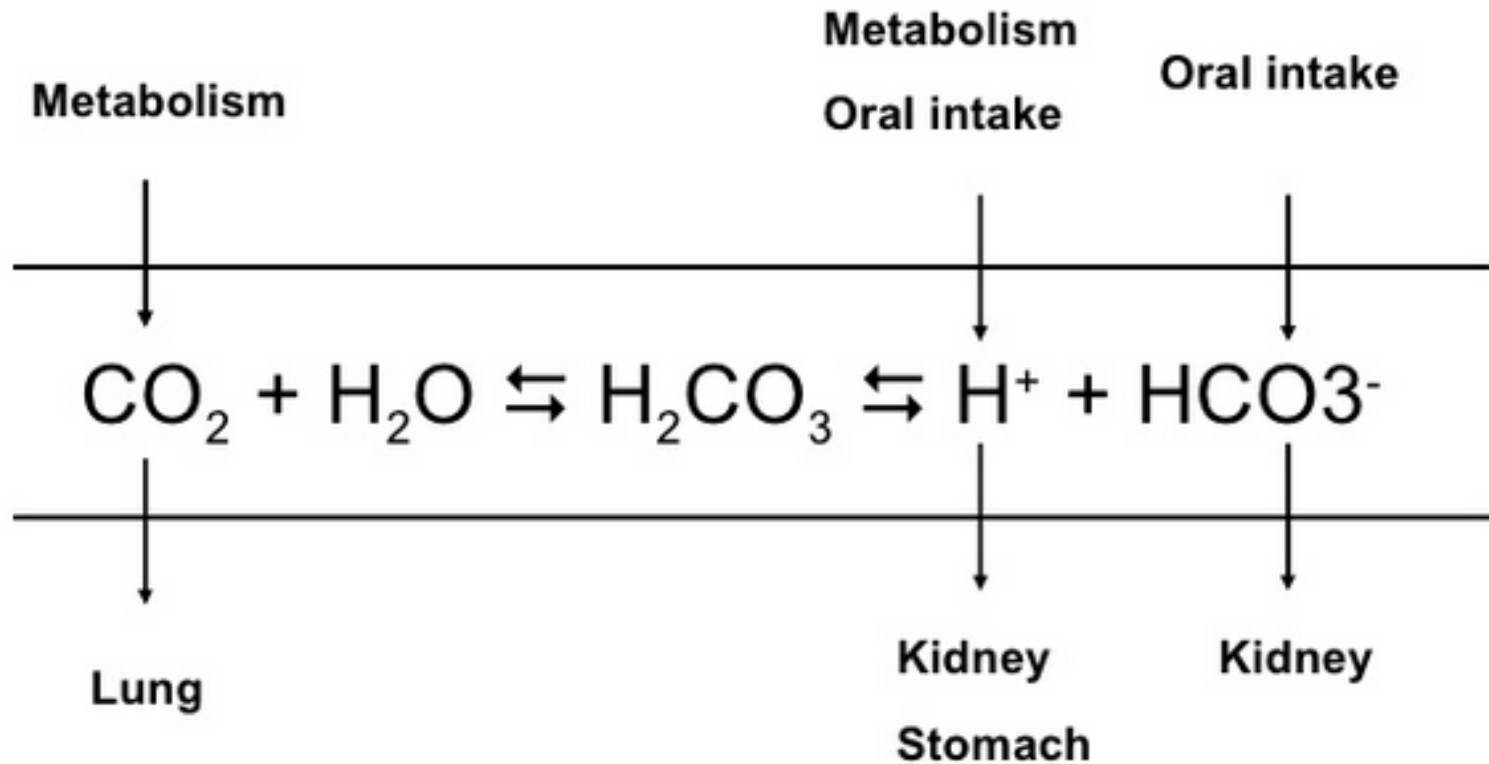
# Compensated or Uncompensated?

1. Evaluate pH—is it normal? Yes
2. Next evaluate pCO<sub>2</sub> & HCO<sub>3</sub>
  - pH normal + increased pCO<sub>2</sub> + increased HCO<sub>3</sub> = compensated respiratory acidosis
  - pH normal + decreased HCO<sub>3</sub> + decreased pCO<sub>2</sub> = compensated metabolic acidosis

# Compensated vs. Uncompensated

1. Is pH normal? No
2. Acidotic vs. Alkalotic
3. Respiratory vs. Metabolic
  - $\text{pH} < 7.30 + \text{pCO}_2 > 50 + \text{normal HCO}_3 = \text{uncompensated respiratory acidosis}$
  - $\text{pH} < 7.30 + \text{HCO}_3 < 18 + \text{normal pCO}_2 = \text{uncompensated metabolic acidosis}$
  - $\text{pH} > 7.50 + \text{pCO}_2 < 30 + \text{normal HCO}_3 = \text{uncompensated respiratory alkalosis}$
  - $\text{pH} > 7.50 + \text{HCO}_3 > 30 + \text{normal pCO}_2 = \text{uncompensated metabolic alkalosis}$

# Buffering



# CO<sub>2</sub>

- ◆ pCO<sub>2</sub> normally is maintained between 35-45 mm Hg. A dissociation curve similar to that for O<sub>2</sub> exists for CO<sub>2</sub> but is nearly linear over the physiologic range of PaCO<sub>2</sub>. Abnormal PCO<sub>2</sub> is always linked to disorders of ventilation and is always associated with acid-base changes.
- ◆ Hypercapnia = Hypoventilation
- ◆ Hypocapnia = Hyperventilation

# Anion Gap

- ◆ Anion gap - this is useful in any cause of metabolic acidosis.
- ◆ In plasma, the sum of the cations is normally greater than that of the anions by approximately 14 mmol/L (normal range 10-18 mmol/L).
- ◆  $(\text{Na}^+ + \text{K}^+) - (\text{Cl}^- + \text{HCO}_3^-) = \text{Anion Gap}$

# Assessing Oxygenation

- ◆ Hypoxaemia-  $paO_2$  less than 60 mmHg
- ◆ Affected by age.
- ◆ Normal  $SaO_2$ 
  - ◆ Arterial: 97%
  - ◆ Venous: 75%



# Important points for assessing tissue oxygenation

- ◆ This is the O<sub>2</sub> that is really available at the tissue level.
- ◆ Is the total Hb normal?
  - ◆ Low Hb means the ability of the blood to carry the O<sub>2</sub> to the tissues is decreased
- ◆ Is perfusion normal?
  - ◆ Low perfusion means the blood isn't even getting to the tissues

# Oxygen Therapy

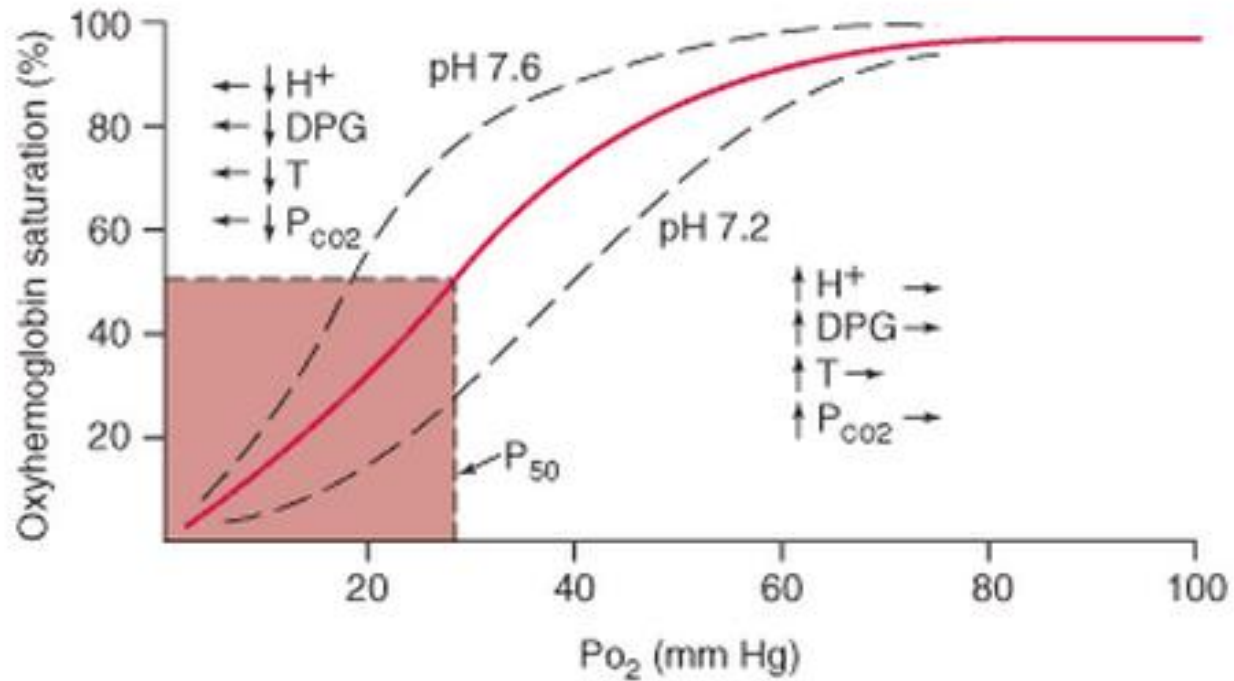
- ◆ Fraction of Inspired Oxygen (FiO<sub>2</sub>): This is the percent of oxygen a patient is inhaling. Room air FiO<sub>2</sub> is 21%. By applying supplemental oxygen, the FiO<sub>2</sub> can go as high as 100%.
- ◆ Est. of % of O<sub>2</sub> involved in exchange at alveoli.
- ◆ Indications for Oxygen Therapy:
  - ◆ To correct hypoxemia
  - ◆ To reduce oxygen demand on the heart
  - ◆ Suspected or acute myocardial infarction (MI)
  - ◆ Severe trauma
  - ◆ Post anesthesia recovery

# Oxygen Therapy

- ◆ Nasal Cannula: The nasal cannula is the most common oxygen device used and the most convenient for the patient. A nasal cannula at 2lpm is usually a good place to start.
- ◆ You may at times need to estimate the  $FiO_2$ . How to estimate  $FiO_2$  on a nasal cannula? For every liter per minute, the  $FiO_2$  increases by 4%.

# Haemoglobin

Oxyhemoglobin dissociation curve.



# Causes of Acidosis

## ◆ Respiratory

- ◆ Hypoventilation
- ◆ Impaired gas exchange

## ◆ Metabolic

- ◆ Ketoacidosis
  - ◆ Diabetes
- ◆ Renal Tubular Acidosis
  - ◆ Renal Failure
- ◆ Lactic Acidosis
  - ◆ Decreased perfusion
  - ◆ Severe hypoxemia

# Causes of Alkalosis

## ◆ Respiratory

- ◆ Hyperventilation due to:
  - ◆ Hypoxemia
  - ◆ Metabolic acidosis
  - ◆ Neurologic
    - ◆ Lesions
    - ◆ Trauma
    - ◆ Infection

## ◆ Metabolic

- ◆ Hypokalemia
- ◆ Gastric suction or vomiting
- ◆ Hypochloremia



# Worked Examples

## 12 year old diabetic presents with Kussmaul breathing

- ◆ pH : 7.05
- ◆ pCO<sub>2</sub>: 12 mmHg
- ◆ pO<sub>2</sub>: 108 mmHg
- ◆ HCO<sub>3</sub>: 5 mEq/L
- ◆ BE: -30 mEq/L



17 year old w/severe kyphoscoliosis, admitted for pneumonia

- pH: 7.37
- pCO<sub>2</sub>: 25 mmHg
- pO<sub>2</sub>: 60 mmHg
- HCO<sub>3</sub>: 14 mEq/L
- BE : -7 mEq/L

9 year old w/hx of asthma, audibly wheezing x 1 week, has not slept in 2 nights; presents sitting up and using accessory muscles to breath w/audible wheezes

💧 pH: 7.51

💧 pCO<sub>2</sub>: 25 mmHg

💧 pO<sub>2</sub> 35 mmHg

💧 HCO<sub>3</sub>: 22 mEq/L

💧 BE: -2 mEq/L

# 7 year old post op presenting with chills, fever and hypotension

- pH: 7.25
- pCO<sub>2</sub>: 32 mmHg
- pO<sub>2</sub>: 55 mmHg
- HCO<sub>3</sub>: 10 mEq/L
- BE: -15 mEq/L

# What Leads to What?

- ◆ PE
- ◆ Vomiting
- ◆ Severe diarrhea
- ◆ Cirrhosis
- ◆ Renal failure
- ◆ Sepsis
- ◆ Pregnancy
- ◆ Diuretic use
- ◆ COPD
- ◆ Metabolic alkalosis
- ◆ Metabolic acidosis
- ◆ Respiratory alkalosis
- ◆ Metabolic acidosis
- ◆ Met. Acid/ Resp. Alk
- ◆ Respiratory alkalosis
- ◆ Metabolic alkalosis
- ◆ Respiratory acidosis